# **COMBUSTION DIAGNOSTICS**

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# PERFORMANCE METRICS

There are three basic performance metrics for a gas turbine combustor

• Fuel air mixing

Fuel air mixing is often quantified by <u>pattern factors</u> which is a measure of <u>uniformity</u> <u>of the temperature distribution</u>. The lower the pattern factor the better the fuel air mixing.

• Volumetric heat release

Higher volumetric heat release enables the design of more compact combustors. Volumetric heat release is also closely related with mixing.

• Emissions (NOx, CO, UHC etc.)

Reducing emissions in order to meet legal criteria







\* Figures courtesy of Siemens-Westinghouse

# ACTIVE CONTROL OF SIDE AIR JETS







**Inside View of the 60 kW Experimental Can Combustor** 

A Siemens-Westinghouse Can **Annular Combustor** 







**Pressure & Heat Release Spectra** 







c. Pixelwise Standard Deviation

# **CH Radical Images**

St = 0



2



3-D Temperature Profiles as a Function of Forcing Frequency

Φ=0.9, R=15





Φ=0.9, R=10









## Emissions Index as a Function of Forcing Frequency

<u>Definition</u>: Emissions index is defined as the grams of nitric oxide generated per kilogram of fuel burnt

Note: Emissions measurements were made using K1 grade kerosene as liquid fuel.

K1 grade kerosene is very close to Jet-A fuel in chemical composition.





### Emissions Index as a Function of Blowing Ratio





Temperature Contours at R=7.8 and Φ=0.87 (Temperatures in °C) **Planar Pattern Factor as a Function of** 

Elevation at R=7.8 and  $\Phi$ =0.87





Synthesis gas (syngas) is a variable mixture of H2 and CO and other some gases in trace amounts

Mostly used in combined cycles (IGCC) to increase efficiency Can be produced from the gasification of coal or biomass Composition depends on the gasification process



•Emissions

# Motivation for Work

- Present low emissions technology mainly focuses on burning natural gas
- ➢Natural gas cannot be relied upon as the exclusive source for fueling the clean power plants in the future
- ➢Increasing demand due to new installations has caused substantial price volatility
- Concerns about future supplies
- Concern about energy security have motivated interest in utilizing coal-derived syngas or fuels from other sources, such as biomass, landfill gas, process gas and others.
- ➢ Technologies like integrated gasification combined cycle (IGCC) plants enable the combustion of coal and other solid or liquid fuels, while still maintaining aggressive emissions targets and high efficiency





# Reasons Causing Flame Flashback

- Low local speeds in the premixing section
- Boundary layer propagation
- Thermo-acoustic instability
- Combustion induced vortex breakdown (CIVB)

• Challenge with syngas is <u>high flame speeds</u> associated with the <u>hydrogen</u> content of the mixture





Schematic View of Fuel Delivery System



# HYDROGEN ENRICHED COMBUSTION STUDIES







Relationship Between Blowout Equivalence Ratio and Reactor Based Loading Parameter

$$S_{L} = \frac{C_{F}}{C_{F} + C_{H}} S_{L,CH_{4},\phi} + \frac{C_{H}}{C_{F} + C_{H}} S_{L,H_{2},s}$$





#### Relationship Between Blowout Equivalence Ratio and Flamelet Based Loading Parameter





Pressure, Heat Release & Flashback Spectra









Qair = 5.7 l/s,  $\Phi$  = 0.7, 40% Hydrogen by Volume

Phase averaged CH radical images with Flashback



#### **Flamelet Modeling**



#### **Combustor Acoustics**

$$\frac{\partial^2 P}{\partial t^2} + c^2 \nabla^2 P = (\gamma - 1) \frac{\partial q'}{\partial t}$$

Acoustic field inside the gas turbine combustor can be assumed to be one-dimensional (i.e. wavelength of acoustic waves much larger than combustor diameter)

$$= \int_{0}^{\infty} \psi^{2}(x) dx$$

L

A sudden jump occurs in properties before and after the flame such that,

$$T(x) = \frac{?T_{in}}{?T_{ad}} \quad x ? x_{f}$$

$$c(x) = \frac{?c_{u}}{?c_{d}} \quad x ? x_{f}$$
Associated change in the speed of sound

Therefore while solving for the mode shape one must consider the matching condition the interface

$$\psi(x_{f}^{+}) = \psi(x_{f}^{-})$$

$$\psi(x) = \begin{cases} \cos\left(\frac{\omega x}{c_{u}}\right) & x \le x_{f} \\ \frac{\cos(\alpha)}{\sin(\beta)}\sin\left(\frac{\varpi(L-x)}{c_{d}}\right) & x > x_{f} \end{cases}$$

Acoustic Mode Shape Along the Length of the Combustor

0.5 **x/L** 



Theoretical Experimental

0.75

Solve for frequency w from the roots of,

$$\tan \alpha \tan \beta = (\rho_u c_u) / (\rho_d c_d)$$





### Heat Release Dynamics

In order to close the thermo-acoustic feedback loop one needs to specify the heat release dynamics

Linear heat release model (Annaswamy et. al., 2002) derived from linearizing a well-stirred reactor model for the flame front with single step chemical kinetics

$$\dot{q}'(s) = \frac{\beta}{s+\alpha} \dot{m}'(s)$$
 Response of heat release fluctuations to mass flow rate perturbations (due to velocity fluctuations)

This linear model predicts linear stability characteristics of the flame but will not yield in a non-linear limit cycle. Instead for unstable conditions system output grows to infinity without an upper bound.

However, it is experimentally observed that heat release saturates as the amplitude of input grows. One can incorporate that non-linear saturation effect as follows,

 $\dot{q}'(s) = \frac{\beta}{s+\alpha} \dot{m}'(s) \left( 1 - \left| \frac{\dot{m}'(s)}{\overline{m}} \right| \right)$  Non-linear response of heat release fluctuations to mass flow rate perturbations

The amount of heat release must be bounded by the amount of fuel available to be burnt. Heat release saturation occurs due to this fuel consumption effect





Flame Front Movement with Respect to Thermo-acoustic Instability Cvcle Position of the Flame front Phase Locked with Pressure



RMS Amplitude of Limit Cycle Pressure Fluctuations

Obtained From Simulations as a Function of Air Flow Rate and Equivalence Ratio (Pure Methane, x/L=0.80)



RMS Amplitude of Limit Cycle Pressure Fluctuations

Obtained From Experiments as a Function of Air Flow Rate and Equivalence Ratio (Pure Methane, x/L=

![](_page_33_Picture_6.jpeg)

![](_page_34_Figure_0.jpeg)

Effect of Hydrogen Volume Fraction and Equivalence Ratio on Dominant Frequency and RMS Pressure Level at a Fixed Air Flow Rate Qair=6.2 l/s

![](_page_34_Picture_2.jpeg)

![](_page_35_Figure_0.jpeg)

**Scaling of RMS Pressure Level** 

![](_page_35_Picture_2.jpeg)

![](_page_36_Picture_0.jpeg)

OH\* Chemiluminesence Images Demonstrating Effect of Fuel Composition on the Distance to the Flame Center of Mass at a Fixed Flow Rate and Equivalence Ratio  $(Qair=6.2 \text{ l/s}, \Phi=0.7)$ 

![](_page_36_Picture_2.jpeg)

![](_page_37_Figure_0.jpeg)

![](_page_37_Picture_1.jpeg)

![](_page_38_Figure_0.jpeg)

Relationship Between Adiabatic Flame Temperature and Emissions Index

![](_page_39_Figure_0.jpeg)

![](_page_40_Figure_0.jpeg)

![](_page_40_Picture_1.jpeg)

![](_page_41_Figure_0.jpeg)

#### Pure Methane

50% H<sub>2</sub> by volume

Phase Locked OH-PLIF Images  $(Q_{air}=6.2 \text{ l/s}, \phi=0.8)$ 

![](_page_42_Figure_0.jpeg)

#### Experimental Setup for PIV

![](_page_42_Picture_2.jpeg)

Cold Flow

![](_page_42_Picture_4.jpeg)

#### Reacting Flow

![](_page_43_Figure_0.jpeg)

In Plane Velocity Components

Swirl Velocity

 $(Q_{air}=6.2 \text{ l/s}, \phi=0.7)$ 

# CONCLUSIONS

• An experimental setup has been developed for studying synthesis gas combustion with a focus on hydrogen enrichment studies

- A mathematical model has been developed
- Hydrogen addition considerably extends lean flammability limits
- Very lean combustion enables near zero nitric oxide emissions
- Wobbe index (as often used with gaseous fuel mixtures) is not a good scaling parameter for hydrogen/methane mixtures
- Hydrogen rich mixtures pose a significant flame holding problem and can also induce a sudden change in the dominant thermo-acoustic mode
- Flashback is primarily triggered by thermo-acoustic instability
- Some passive strategies employed for flashback control (i.e. changes in injector & pre-mixer geometries) , however active control strategies may be employed as well
- Several blowout mechanisms have been examined

![](_page_44_Picture_10.jpeg)

![](_page_45_Picture_0.jpeg)