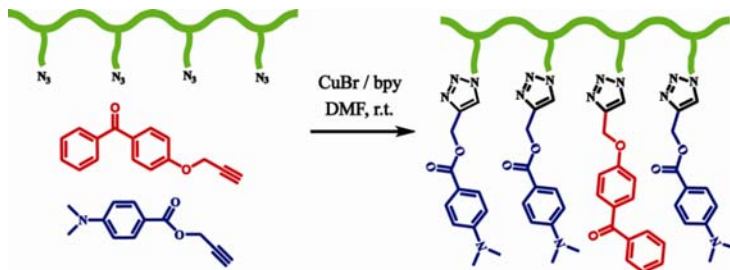


Synthesis and Characterization of One-Component Polymeric Photoinitiator by Simultaneous Double Click Reactions and Its Use in Photoinduced Free Radical Polymerization

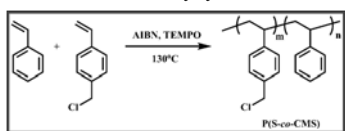
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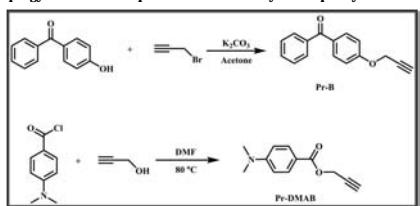
ABSTRACT: A polystyrene copolymer (PS-B-DMAB) possessing both benzophenone and dimethylamino moieties in the side chain was synthesized by a combination of nitroxide-mediated radical polymerization (NMRP) and simultaneous double “click” reactions. First, a random copolymer of styrene (S) and chloromethylstyrene (CMS) with 32 mol % CMS content was prepared by NMRP process. Then, chloromethyl groups were converted to azide groups by reacting with NaN₃ in DMF. The other two click components, namely, propargyl benzophenone (Pr-B) and propargyl 4-(dimethylamino)benzoate (Pr-DMAB), were prepared independently by the etherification and esterification reactions, respectively. Then, in the final stage, the obtained alkyne functional chromophoric (Pr-B) and hydrogen-donating (Pr-DMAB) molecules were anchored to azide-modified polystyrene (PS-N₃) in one-step by “click chemistry”. The final polymer (PS-B-DMAB) and the intermediates were characterized in detail by spectral analysis and laser flash photolysis studies.



Nitroxide-Mediated Radical Copolymerization of Styrene and Chloromethylstyrene



Propargylation of Benzophenone and Dimethylaminophenyl Derivatives



Synthesis of One-Component Polymeric Photoinitiator by Simultaneous Double Click Reaction

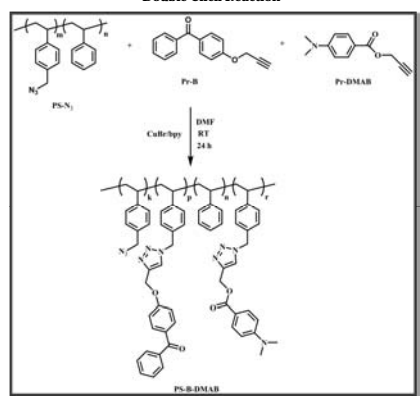


Table 1. Photoinitiated free radical polymerization of MMA at $\lambda_{\max}=350$ nm.^a

Run	Photoinitiator ^b (mol L ⁻¹)	Solvent	$R_p \times 10^5$ (mol L ⁻¹ s ⁻¹)	M_n^c (g mol ⁻¹)
1	B (1x10 ⁻²)	CHCl ₃	-	-
2 ^d	B (1x10 ⁻²)	CHCl ₃	13.63	17350
3	PS-B-DMAB (1x10 ⁻²)	CHCl ₃	4.99	50900
4	PS-B-DMAB (1x10 ⁻²)	CHCl ₃ ^e	7.14	111300
5	PS-B-DMAB (2x10 ⁻³)	THF	5.52	45790
6	PS-B-DMAB (2x10 ⁻³)	DMF	7.14	90500

^a [MMA] = 4.68 mol L⁻¹, Time = 2 h.

^b The photoinitiator concentration is given in terms of benzophenone moieties.

^c Determined by GPC according to linear polystyrene standards.

^d Benzophenone/ N,N-dimethylaniline: 1/ 3 is used as photoinitiating system.

^e Carried out under nitrogen atmosphere.

Figure 5. ¹H NMR spectra of PS-B-DMAB and poly(styrene-graft-methyl methacrylate) (P(S-g-MMA)), obtained after photolysis in the presence of MMA.

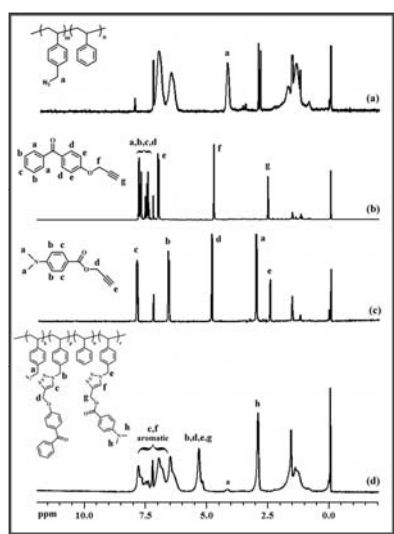


Figure 1. ¹H NMR spectra of a) PS-N₃, b) Pr-B, c) Pr-DMAB, and d) PS-B-DMAB in CDCl₃.

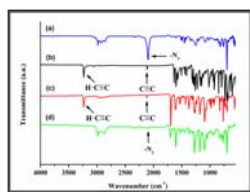


Figure 2. FT-IR spectra of a) PS-N₃, b) Pr-B, c) Pr-DMAB, and d) PS-B-DMAB.

Photoinitiated Free Radical Polymerization of Methyl Methacrylate (MMA) by Using Polymeric Photoinitiator (PS-B-DMAB)

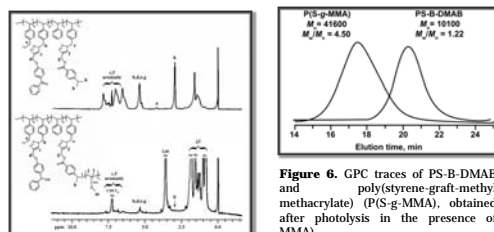
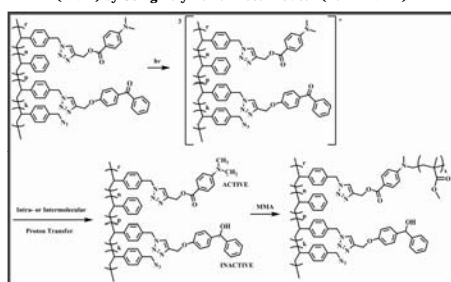


Figure 6. GPC traces of PS-B-DMAB and poly(styrene-graft-methyl methacrylate) (P(S-g-MMA)), obtained after photolysis in the presence of MMA.

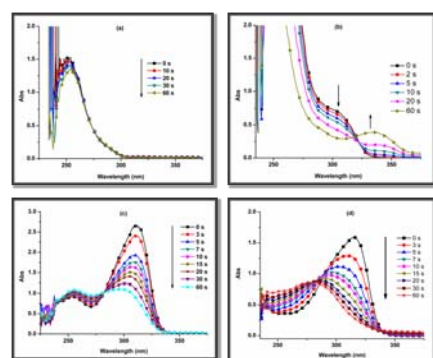


Figure 3. Absorption spectra of a) B (2.5×10^{-5} mol L⁻¹), b) B (2.5×10^{-5} mol L⁻¹)/DMA (7.5×10^{-5} mol L⁻¹) c) B (2.5×10^{-5} mol L⁻¹)/EDAB (7.5×10^{-5} mol L⁻¹) and d) PS-B-DMAB (1.3×10^{-5} mol L⁻¹, in terms of benzophenone moieties) in CHCl₃.

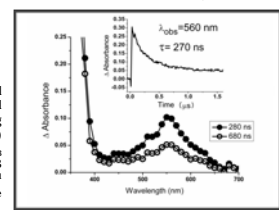


Figure 4. Transient optical absorption spectrum recorded at 280 and 680 nm following laser excitation (355 nm, 5 ns) in nitrogen saturated CHCl₃ solution of PS-B-DMAB (concentration = 2×10^{-5} mol L⁻¹ in terms of benzophenone moieties).

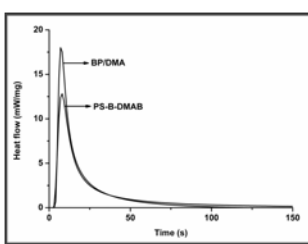


Figure 7. Heat flow vs. time for the polymerization of TMPTA initiated by PS-B-DMAB and BP/DMA systems, cured at 30°C by UV light with an intensity of 53 mW cm⁻². (The photoinitiator concentration is 0.0015 M in terms of BP moieties.)

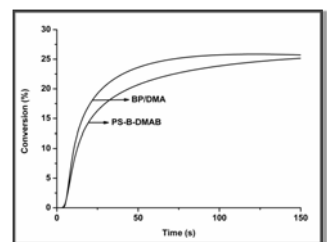


Figure 8. Conversion vs. time for the polymerization of TMPTA initiated by PS-B-DMAB and BP/DMA systems, cured at 30°C by UV light with an intensity of 53 mW cm⁻². (The photoinitiator concentration is 0.0015 M in terms of BP moieties.)

References

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