

**Drop your homework to assistant's (M.B. Kelebek) mail box or office desk.**

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**Related Lecture: Lecture notes on Solar and Atmospheric Radiation before you answer.**

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**Q.1. (a)** Knowing that the solar constant ( $I_0$ ) for Earth is  $1368 \text{ W/m}^2$ , and that the solar radius  $6.9 \times 10^8$  [m], derive an equation for the effective surface temperature of the Sun and show that it is about  $6000^\circ\text{K}$ .

**(b)** At what wavelength will be the maximum energy radiated from the Earth's ground if its surface temperature is  $300$  [ $^\circ\text{K}$ ]. Compare this with the maximum wavelength that Sun radiates at  $5870 \text{ K}$ . Which object will radiate more energy?

**(c)** Make a sketch of the electromagnetic spectrum (Planck spectrum) for each object radiating at the temperatures given in (b). Indicate the spectral regions on your plot. Do not make separate plots for each object. Show them all in ONE graph.

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**Q.2. (a)** Assuming the surface temperature of the Sun is  $5870^\circ\text{K}$ , calculate the total solar radiation per unit area per unit time at the top of the atmospheres of Earth, Mars, and Jupiter. (1 AU =  $1.5 \times 10^8 \text{ km}$ , Mars distance is 1.5 AU, Jupiter's distance from Sun is 5 AU).

**(b)** Neglect the effects of the atmospheres of these and calculate the equilibrium or effective temperature ( $T_e$ ) for each planet.

**(c)** **Make a Table** that shows your results of steps in (a, and b). See Table below as an example.

**(d)** Compare your results with the Observed Mean temperature given for these planets as  $288^\circ\text{K}$ ,  $210^\circ\text{K}$ ,  $150^\circ\text{K}$ . Assume that the dominant gas in the atmospheres of these planets are air,  $\text{CO}_2$  and ammonia, **respectively**. Using this information, explain why the calculated and observed temperatures are different.

Use the planetary data sheet on our web page for the constants you need in this part.

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**Q.3. Brain Storm / Search and Learn:** In this part, you are expected to search and learn yourself.

**a.** Explain the differences between Planck Law and Stephan-Boltzmann Law?

**b.** Explain the differences between temperature and heat? What are their units?

**c.** What is Albedo? Explain in which case a planet receives more energy at the surface: a planet with albedo or without albedo? Why?

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\* **Note:**

1. Homework returned after due date will not be accepted.
2. Electronically typed Homework is not accepted. Return your homework in paper work with your handwriting.
3. Photocopied or Scanned homework is not accepted.
4. **Do not use COMPUTER PRINTER OUTPUTS** for your homework unless it is with **your handwriting**.